Department of Chemistry. Course objectives and Outcomes

F.Y.B.Sc.

SemI		
Paper	Objectives	Outcome
CH-101 Physical and Inorganic Chemistry	Develop the knowledge of basic principles of physical chemistry and to introduce mathematical preparations. Introduction to periodic properties with reference to S-Block elements	Students are able to understand the basic principles of physical chemistry and the skills of mathematical preparations. Students understood periodic properties of S-Block elements and were able to co-relate the same.
CH-102 Organic and Inorganic Chemistry	Give introduction of organic chemistry with reference to hydrocarbon, halogen derivatives, alcohol phenol ether. Knowledge of ionic Equiliberia, Knowledge of shapes of covalent molecules with reference to VSEPR.	Students understood organic chemistry with reference to hydrocarbon, halogen derivatives, alcohol phenol ether. Students came to know various terms, equations and ability to solve numerical of ionic Equiliberia, They learnt to draw shapes of electronic structures of covalent molecules.
CH-103 Chemistry Practical	Develop the skill to handle instruments, preparation of solutions and analytical experiments. Introduction to basic inorganic qualitative analysis.	Students were able to develop the skill to handle instruments, preparation of solutions and analytical experiments. Students understood basic inorganic qualitative analysis.
SemII		
CH-201 Physical and	Develop knowledge of	Students were able to understand

Inorganic Chemistry	physical properties of	physical states of matter.
morganic chemistry	matter.	They are able to understand
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	Develop knowledge of	metallurgical processes and properties
	metals and metallurgy as	of P-Block
	well as P-Block elements	
CH-202 Organic and	Introduction to various	Students gained the knowledge of
Inorganic Chemistry	organic compounds like	various organic compounds like
	aldehydes, ketones,	aldehydes, ketones, carboxylic acids
	carboxylic acids and	and their derivatives with reference to
	their derivatives.	their preparation and properties.
	Basic knowledge of	Students understood volumetric
	volumetric analysis and	analysis and bond and structure of
	bond and structure of	molecules.
	molecules.	
CH-203	Develop skill to perform	Students were able to develop skill to
Chemistry Practical	simple physical	perform simple physical chemistry
	chemistry experiments	experiments and analytical chemistry
	and analytical chemistry	experiments.
	experiments.	Students were introduced to basic
	Introduction to basic	organic qualitative analysis.
	organic qualitative	
	analysis.	

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SemIII		
CH-301	To know Physical	Students came to know the physical
Physical and Inorganic	properties of solutions,	properties of solutions, and understand
chemistry	colligative properties of	colligative properties and able to solve
	solution.	the numerical.
	Introduction to D-Block	They understood various terms of the
	elements.	D-Block Elements.
CH-302 Organic and	Introduction to	Students gained the knowledge of
Inorganic Chemistry	stereochemistry with	stereochemistry with reference to
	reference to projection	projection formula, optical and
	formula, optical and	geometrical isomers, conformational
	geometrical isomers,	isomers, stereochemistry of cyclohexan.
	conformational isomers,	They developed knowledge of
	stereochemistry of	heterocyclic and polycyclic compounds,
	cyclohexan.	solvents, solutions, acids and bases.

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	To develop knowledge	
	of heterocyclic and	
	polycyclic compounds,	
	solvents, solutions, acids	
	and bases.	
CH-303	Develop skill to perform	Students are able to perform practical
Chemistry Practical	physical chemistry	based on physical and organic
	experiment, volumetric	chemistry, chromatography and
	and chromatographic	volumetric analysis.
	analysis.	
	Prepare organic	
	compounds.	
CH-304	Introduction to analytical	Students were able to understand
Advanced Analytical	chemistry and	principle and of acid base titration and
Chemistry(Skill	volumetric analysis with	precipitation titration and
enhancement course)	reference to acid base	chromatographic methods.
	titration and precipitation	Students were able to understand
	titration and	sampling, accuracy, errors and good
	chromatographic	laboratory practices.
	methods.	
SemIV		
CH-401	To give knowledge of	Students gained the knowledge of
Physical and Inorganic	electro chemistry and	electrochemistry and chemical
chemistry	chemical	thermodynamics and gained the ability
	thermodynamics.	to solve the numerical.
	Introduction to	Students understood coordination
	coordination chemistry.	chemistry, complexes, their
		nomenclature, Ligands and chelates
CH-402	Introduction to organic	Students knew synthetic reagents and
Organic and Inorganic	synthesis with reference	organo metallic compounds, their
Chemistry	to AAE, ME and organo	preparation and uses.
	metallic compounds.	Students developed the ability to
	Introduction to	understand combination of orbitals,
	molecular orbital theory	molecular orbital treatment LCAO
CH-403	To develop skill to	Students developed skill to handle
Chemistry Practical	handle instruments and	instruments and perform physical
	perform physical	chemistry experiments.
	chemistry experiments.	Students learnt to identify organic
	To identify organic	compounds, to prepare inorganic
	compounds, to prepare	compounds and gravimetric analysis
	inorganic compounds	

	and gravimetric analysis	
CH-404 Advanced Analytical Chemistry(Skill enhansment course)	To know the volumetric analysis with reference to redox and complexometric titration, methods and gravimetric analysis.	Students gained the knowledge of the volumetric analysis with reference to redox and complexometric titration, methods and gravimetric analysis.

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SemV		
CH-501 - Principles of	To acquire knowledge	After successful completion of this
Physical Chemistry-I	about rates of chemical	course, students are expected to:
	reactions and	Understand the significance of wave
	distinguishing the reaction of different	function and postulates of quantum mechanics.
	order and their	Deduce rate equations and half-life
	characteristics.	equations for first and second order
	To understand the basic	reactions
	principles of phase rules	Draw and explain the one and two
	and phase diagrams.	component system phase diagrams.
	To learn the underlying	Explain the principles of electrode
	principles of electrode	processes and apply them during
	reactions, electrochemical	Practicals.
	cells and applications of EMF	
CH-502 Subject-	To describe the VSEPR	Learn about the VSEPR theory and
Inorganic Chemistry	theory to predict shape	how it can be used to explain molecular
	of molecules from	shapes.
	electron pairs.	Learn about the VBT to describe the
	To describe the bonding	formation of covalent bonds in terms of
	in simple compounds	atomic orbital overlap.
	using VBT.	Learn about stability of complexes
	To describe the	using CFSE.
	principles of VBT to	Learn about MOT to draw energy
	predict hybridization of	diagrams and to predict bond order
	orbitals.	
	To understand how CFT	
	explains electronic	

	atmastra	
	structure, colour and	
	magnetic properties of	
	co-ordination	
	compounds.	
	To introduce the basic	
	principles of MOT and	
	electronic geometry of	
	molecules	
CH-503- Organic	To study different types	Students will learn organic reactions
Reaction Mechanism	of organic reactions.	like nucleophilic substitution,
	To understand the	electrophilic substitution, nucleophilic
	mechanisms of different	addition, electrophilic addition and
		elimination.
	types of reactions.	
	To distinguish between	Students will be able to write/ explain
	types of substrates and	mechanisms of those types of reactions.
	types of reagents.	Students will understand how a reaction
	To understand ways of	takes place in one or more steps.
	attack of reagent,	Students will understand the types of
	breaking and formation	intermediates formed in different
	of bonds in different	reactions.
	reaction mechanisms.	Students will learn how reagent attacks
	To study kinetics,	the substrate molecule and accordingly
	evidences and factors	how bonds break and formed.
	affecting different types	Students will learn how change in
	of reactions.	structure of substrate, reagent and
	To study stereochemistry	solvent changes the product formed and
	of different reactions.	its stereochemistry.
	To understand role of	Students will be able to predict the
	different reagents in	products and to suggest the mechanisms
	different reactions	
CH-504 - Industrial	To produce graduates	Student will be able to understand:
Chemistry	with enhanced skills,	Basic requirements of Chemical
	applied knowledge,	Industry, different terms, operations and
	aptitude to carry out	processes involved in chemical
	higher studies or	Industry.
	research and	Describe Copy Right Act, Patent Act
	development in the	and Trade Marks, Bureau of Indian
	various industrial areas.	Standards (BIS) and International
	To make the student	Organization for Standardization (ISO).
	cognizant about	Basic requirements, raw materials,
	important aspects of	different processes and operations
	Chemical Industries,	involved in Sugar Industry and also
	Industrial work culture	different grades of sugar and uses of
		unrerent graves of sugar and uses of

	and anning sector	her man denotes of an and in the state
	and environment.	by-products of sugar industry.
	To prepare the students	Importance of fermented products,
	for immediate entry to	basic requirements, theory and process
	the workplace with	of alcohol making, fractional
	sound theoretical	distillation and various terms involved
	knowledge and some	in Fermentation Industry.
	basic experimental	Understand Occurrence of Petroleum,
	concepts in the area of	theories of formation of Petroleum and
	various industries viz.	different terms Viz. Knocking, Anti-
	Sugar Industry,	Knock Compounds, Octane number,
	Fermentation Industry,	Cetane number, Gasohol and Power
	Petroleum and	alcohol etc.
	Petrochemicals.	Manufacturing processes involved in
	To offers the synergism	Industrial Organic Synthesis such as
	between basic concepts	Methanol, Isopropanol, Glycerol,
	of Chemistry with	Acetylene and Aromatic hydrocarbon
	Industrial applications.	i.e. Toluene from petroleum with their
	To equip the students	uses.
	with knowledge of some	
	industrial organic	
	synthesis as requirement	
	of diverse chemical	
	industries.	
	Empower the students to	
	understand the concepts	
	in chemical	
CH-505 Subject-	To develop an	Explain the fundamentals of analytical
Analytical	understanding of the	methods and instruments for qualitative
Instrumentation	range and uses of	and quantitative Analysis.
	analytical methods in	Express the role of analytical chemistry
	chemistry.	in science.
	To understand and	Students will be able to function as a
	establish the role of	member of aninterdisciplinary problem
	chemistry in quantitative	solving team.
	analysis.	
	To enhance the	
	Analytical instrumental	
	skill of the students.	
CH-506(A) -	To study different types	Students will study biomolecules like
Biochemistry	of biomolecules.	carbohydrates, amino acids, proteins,
	To study structure of	enzymes, lipids and nucleic acids.
	biomolecules.	Students will understand definitions,
	To study classification of	classifications and examples of these

	angle type of	hismologylag
	each type of	biomolecules.
	biomolecules.	Students will learn the detailed
	To study reactions of the	structure of these biomolecules along
	biomolecules.	with types of bonds or linkages present
	Study of metabolism and	in their molecules.
	thus, study of metabolic	Students will learn the chemical
	processes and reactions	properties of these biomolecules and
	involved.	the action of some reagents on them in
	To study energetics of	the form of reactions or graphical
	the metabolic processes.	presentation.
	Students should	Students will understand biochemical
	understand: Structure	energetics of common energy rich
	and role of	compounds along with hydrolytic
	Carbohydrates, Amino	reactions. Students will learn
	acids, Proteins,	metabolisms like Glycolysis, TCA
	Enzymes, lipids, Nucleic	cycle, Transamination, deamination and
	Acids and energy rich	β - oxidation through reactions, enzymes
	compounds in	involved, outlines and energetics
	biochemical reactions	
CH-507 Physical	To develop skills	Students will get basic analytical and
Chemistry Practical	required in chemistry	technical skills to work effectively in
5	such as the appropriate	the various fields of chemistry.
	handling of apparatus,	Students will able to calibrate and
	instruments and	handle instruments like conductometer,
	chemicals.	potentiometer, pH meter, colorimeter,
	The student will learn	spectrophotometer, polarimeter.
	the laboratory skills	They have ability to perform accurate
	needed to design, safely	quantitative measurements with an
	conduct and interpret	understanding of the theory and use of
	chemical research.	contemporary chemical
	To expose the students to	instrumentation, interpret experimental
	an extent of	results, perform calculations on these
	experimental techniques	results and draw reasonable, accurate
	using modern	conclusions.
	instrumentation.	They get skills required in chemistry
	The student will develop	such as the proper handling of
	-	apparatus and chemicals.
	the ability to effectively communicate scientific	
	information and research	They will have ability to present scientific and technical information
	results in written and	resulting from laboratory
	oral formats	experimentation in both written and
		oral formats. Students will apply
		conductometer, potentiometer, pH

		meter, colorimeter, spectrophotometer, polarimetery techniques for analysis and measurement.
CH-508 -Inorganic	To analyze the inorganic	Student will able to determine cation &
Chemistry Practical	mixtures.	anion from inorganic mixtures by using
	To determine metal from	qualitative analysis.
	ore and alloy analysis.	Student will able to determine metal
	Using colorimetric	from ore & alloys.
	analysis to determine	Students will be able to design & carry
	amount of metal.	out scientific experiments as well as
		accurately record & analyze the results
		of experiments.
		Students will be able to handle
		colorimeter for estimation of metal ions
CH-509 - Organic	To develop skills	Separate and analyze binary water
Chemistry Practical	required in chemistry	insoluble mixture.
	such as the appropriate	Separate and analyze binary water
	handling of apparatus	soluble mixture.
	and chemicals.	Estimate - Acetamide, Glucose and
	The student will learn	Glycine by volumetric method,
	the laboratory skills	Estimate basicity of various acids.
	needed to design, safely	Synthesis of various organic
	conduct and interpret	compounds through greener
	chemical research.	alternatives.
	To expose the students to	Understand Thin Layer
	an extent of	Chromatographic techniques and
	experimental techniques	physical constant.
	using modern	Understand the purification technique
	instrumentation.	use in organic chemistry.
	The student will develop	
	the ability to effectively	
	communicate scientific	
	information and research	
	results in written and	
	oral formats	
SemVI		
CH-601 - Principles of	To learn the basics of	After successful completion of this
Physical Chemistry-II	molecular spectroscopy	course, students are expected to:
	and rotational spectra.	Analyze the rotational spectra of
	To understand the basic	diatomic molecules and determine the
	principles and	bond length.

	applications of nuclear chemistry. To learn the consequences of light absorption by atoms and molecules and photochemical reactions. To learn the laws of crystallography and basics of crystal structure	Explain and apply the radioactivity principles for various chemical and biological investigations. Describe the mechanism of fluorescence, phosphorescence and photochemical reactions. Analyze the given crystal structure and determine the indices of planes, interplaner distances and type of crystal structure
CH-602 - Chemistry of Inorganic Solids	To describe the VSEPR theory to predict shape of molecules from electron pairs. To describe the bonding in simple compounds using VBT. To describe the principles of VBT to predict hybridization of orbitals. To understand how CFT explains electronic structure, colour and magnetic properties of co-ordination compounds. To introduce the basic principles of MOT and electronic geometry of molecules.	Learn about basic principles and synthesis of nanomaterials. Learn about classification, composition and processing of cement. Learn about classification and composition of alloys. Learn about types manufacture and applications of fertilizers.
CH-603 - Spectroscopic Methods of Structure Determination	To study principle of spectroscopy and to understand wave parameters and terms involved in spectroscopy. To study different types of spectroscopy. To understand principle, concept and the terms used in each type of	Students will learn interaction of radiations with matter. They will understand different regions of electromagnetic radiations. They will know different wave parameters. Students will learn principle of mass spectroscopy, its instrumentation and nature of mass spectrum. Students will understand principle of UV spectroscopy and nature of UV spectrum. They will learn types of

spectroscopy.	electronic excitations.
Interpretation of UV, IR,	Students will be able to calculate
NMR spectra.	maximum wavelength for any
Use of spectral data for	conjugated system. And from the value
determination of	of λ -max they will be able to find out
structure of unknown	extent of conjugation in the compound.
organic compounds.	Students will understand principle of IR
To study different	spectroscopy, types of vibrations and
applications of each type	the nature of IR spectrum. From IR
of spectroscopy	spectrum, they will be able to find out
1 10	IR frequencies of different functional
	groups. And thus, they will be able to
	find out functional groups present in the
	compound.
	Students will understand principle of
	NMR spectroscopy and will understand
	various terms used in NMR
	spectroscopy. They will learn
	measurement of chemical shift and
	coupling constants.
	Students will be able to interpret the
	NMR data and they will be able to use
	it for determination of structure of
	organic compound. Students will be
	able to determine structure of simple
	organic compounds on the basis of
	spectral data such as λ max values, IR
	frequencies, chemical shift (δ values).

CH-604 - Chemistry of	To make student	Student will be able to understand:
Industrially Important	perceptive about various	Describe the industrial production of a
Products	commodity industries	number of important organic and
	viz. Cosmetics and	inorganic compounds / chemicals and
	Perfumes, Dyes and	products of end use.
	Pharmaceuticals,	Gain comprehensive knowledge of
	Pesticides, Soaps and	cutting-edge developments in a field of
	Detergents, related	different chemical industries.
	diversified and	Importance of Cosmetics Industry and a
	multidisciplinary fields	general study including preparation and
	of chemical industry.	uses of the Hair dye, hair spray,
	To produce graduates	shampoo, suntan lotions, lipsticks,
	with enhanced skills,	talcum powder, nail enamel, creams
	knowledge and research	(cold, and shaving creams).
	aptitude to carry out	Perfumes and identify the
	higher studies or	distinguishing features of its
	research and	components and also an essential oils
	development in the	and their importance in cosmetic
	various industrial areas.	industries with reference to Eugenol,
	To equip students with	Geraniol, sandalwood oil, eucalyptus,
	advance knowledge	rose oil, 2- phenyl ethyl alcohol,
	about various	Jasmone, Civetone, Muscone etc.
	industrially important	Know about pesticides both natural and
	products.	synthetic, benefits and adverse effects
	To makes students ready	of it, also synthesis, manufacture and
	for immediate entry to	uses of pesticides viz. Organochlorines
	the workplace with	(DDT, Gammexene,);
	sound theoretical and	Organophosphates (Malathion,
	basic experimental	Parathion); Anilides (Alachlor and
	knowledge in the areas	Butachlor).
	of various industries.	Definition, classification, raw material
	To engender the	used in soaps and detergents, reaction
	substantial interest in the	involved in it, Manufacture of Soaps
	students to understand	and cleansing action of soaps and
	the concepts in chemical	detergents. Definition, properties of
	processing, engineering	good dyes, relation between colour and
	and industrial	constitution, classification of dyes
	development of present	according to their mode of application
	era viz. Cosmetics and	and chemical constitution.
	Perfumes Industry, Dyes	Importance's, definition and meaning
	and Pharmaceuticals,	of the different terms involved in Drugs
	Pesticides, Soaps and	and Pharmaceuticals Industry and also
	Detergents, related	synthesis, uses, properties and

	manifi dia similimana ana 1	inductivial manufacture of Decent 1
	multidisciplinary and	industrial manufacture of Paracetamol,
	diversified fields of	Aspirin, and Chloramphenicol.
	chemical industry.	
	To describe the	
	industrial production of a	
	number of important	
	organic and inorganic	
	compounds / chemicals	
	and products of end use.	
	To gain comprehensive	
	knowledge of cutting-	
	edge developments in a	
	field of different	
	chemical industries by	
	discussions and	
	exchange of experiences	
	and knowledge.	
CH-605 Analytical	To provide knowledge of	Students are able to gain the knowledge
Chemistry	instruments which are	of instruments which are used in
-	used in Chemical,	Chemical, Pharma, Petroleum, and
	Pharma, Petroleum, and	insecticide and pesticide industry
	insecticide and pesticide	Technical skills of students were raised
	industry	as per industry need.
	To increase student	Develop an understanding of the range
	technical skill as per	and uses of analytical methods in
	industry need.	chemistry.
	To develop an	5
	understanding of the	
	range and uses of	
	analytical methods in	
	chemistry	
CH-606(A) - Polymer	The course offers the	Define terms like monomer, polymer,
Chemistry	basic concepts of	polymerization, polydispersity index,
	polymer, polymerization,	etc., classify polymers based on their
	classes of polymers,	origin, native backbone chain, and
	important properties, and	thermal response.
	poly(lactic acid) as a	Know glass transition temperature and
	biodegradable polymer.	its determination, various ways to
	The course also offers to	express molecular weights of polymers
	study preparation,	and polydispersity index.
	properties, and	Identify different mechanisms of
	applications of	polymerizations viz. free radical, ionic,
	industrially important	and condensation polymerizations.
	moustitany important	and condensation porymenzations.

	selected polymers. The course will give chance to study various mechanisms of polymerization and learn different techniques of polymerization. The student will be able to understand glass transition temperature and factors affecting on it and various ways to express molecular weight of polymers.	Distinguish techniques of polymerization based on physical conditions required for the preparation of polymers in laboratory or industry. Familiar with preparation, properties, and applications of industrially important selected polymers.
CH-607 - Physical Chemistry Practical	To develop skills required in chemistry such as the appropriate handling of apparatus, instruments and chemicals. The student will learn the laboratory skills needed to design, safely conduct and interpret chemical research. To expose the students to an extent of experimental techniques using modern instrumentation. The student will develop the ability to effectively communicate scientific information and research results in written and oral formats.	Students will get basic analytical and technical skills to work effectively in the various fields of chemistry. Students will able to calibrate and handle instruments like conductometer, potentiometer, pH meter, colorimeter, spectrophotometer, polarimeter. They have ability to perform accurate quantitative measurements with an understanding of the theory and use of contemporary chemical instrumentation, interpret experimental results, perform calculations on these results and draw reasonable, accurate conclusions. They get skills required in chemistry such as the proper handling of apparatus and chemicals. They will have ability to present scientific and technical information resulting from laboratory experimentation in both written and oral formats
CH-608 - Inorganic Chemistry Practical	To determine metal from gravimetric estimations. To determine amount of metal by volumetric analysis. To determine preparation /synthesis of co-	Students will be able to prepare co- ordination compounds. Students will be able to determine amount of metal by using quantitative analysis. Students will be able to calculate Rf value of metal.

	ordination compound. To study separation techniques of metals. To use colorimetric analysis of metal	Students will be able to design & carry out scientific experiments as well as accurately record & analyze the results of experiments. Students will be able to explain why chemistry is an integral activity for addressing social, economic & environmental problems.
CH-609 - Organic Chemistry Practical	To develop skills required in chemistry such as the appropriate handling of apparatus and chemicals. The student will learn the laboratory skills needed to design, safely conduct and interpret chemical research.	Separate and analyze binary water insoluble mixture. Separate and analyze binary water soluble mixture. Estimate - Acetamide, Glucose and Glycine by volumetric method, Estimate basicity of various acids. Synthesis of various organic compounds through greener alternatives. Understand Thin Layer Chromatographic techniques and physical constant. Understand the purification technique use in organic chemistry

M.Sc.-I

SemI		
CH-110 Physical	Introduction to	Students understood quantum
Chemistry	quantum chemistry,	chemistry, Nuclear and radiation
	nuclear chemistry and	chemistry, Electrochemistry,
	adsorption	Adsorption-principles, rules, theories
		and numerical problems based on this
CH-130 Inorganic	Introduction to MOT,	Introduce to MOT, Organometallic
Chemistry	Organo metallic	compounds, Molecular symmetry-
	compounds, symmetry	elements of symmetry, point groups,
	and group theory and	introduction to transition metals,
	transition metals.	
CH-150 Basic Organic	Stereochemistry,	Introduction and revision of
Chemistry	reaction mechanisms	Stereochemistry, Basic Mechanisms of
		organic reactions like Nucleophilic

		substitution reaction, Electrophilic substitution reaction, Addition and elimination reactions.
SemII		
CH-210 Physical Chemistry	Introduction to thermodynamics, spectroscopy	Thermodynamics, Statistical Thermodynamics, chemical, kinetics Molecular spectroscopy- students understood principles, theories, rules derivations and numerical problems based on this.
CH-230 Inorganic Chemistry,	Introduction to transition metal complex, ionic bonds, catalysis	Students understood Reaction mechanism in transition metal complexes, ionic bonds, Catalysis, spectra, preparation and applications of complexes.
CH-250 Name reaction, Synthetic Organic Chemistry& Spectroscopy	Introduction to some name reactions	Students understood Name reactions, Synthetic reagents, Rearrangement reactions which are useful for organic synthesis, introduce to spectroscopy.
General Chemistry	Introduction to basic analytical chemistry and maths related to chemistry	Introduced to basic analytical chemistry and maths related to chemistry
CH_P_I-Physical chemistry Practical- yearly	Handling of Instruments, perform of experiment, calculation	Develop a skill to handle instruments, preparation of solution and calculations.
CH-I-I- Inorganic chemistry practical- Yearly	Introduction to ore analysis, binary mixture analysis, drug analysis	Able to analyses ores like pyrolusite, Haematite, Chromite, Dolomite. Develop a skill to perform binary mixtures, analysis of drugs, chromatography
CH-O-I Organic Chemistry Practical Yearly	Organic preparations TLC, use of software to draw structures of organic compounds.	Develop a skill to prepare organic compounds in single stage monitored by TLC. Use of software like ISI draw, chem. Draw, Chem. Sketch to design reaction mechanism, IUPAC names etc.

M.Sc.-II

SemIII			
CH-350 Organic Reaction Mechanism	Physical approach to organic chemistry, reaction intermediates, neighbouring group effect, carbon nucleophile reactions, ester hydrolysis.	Students understood Strength of acids and bases, Determining mechanism of a reaction, Intermediated and concerted Reaction, linear free energy relationship, Aromaticity and neighboring group effect. Hydrolysis of ester and reactions of carbon nucleophile	
CH-351Spectroscopic methods in structure determination	Introduction to H1 NMR, C1NMR Mass spectroscopy. Problems related to this	Students introduce to H NMR, C NMR, Mass spectroscopy, and problems based on spectroscopy	
CH-352 Organic Spectroscopy	Introduction to stereochemistry, Asymmetric synthesis, stereochemistry of six member and other then six member ring	Principle of spectroscopy, Asymmetric synthesis and applications, Stereochemistry of six member ring, other than six member ring, fused rings.	
CH-353Free radical, photochemistry, pericyclic reactions and their applications	Introduction to free radical and photochemistry, aromatic compound alkenes. Pericycle reactions, electrocyclic reactions	Students understood quantum yield and electronic state. Norrish –I and Norrish –Iicharges. Paterno-Buchi reaction Photochemistry of olifines and arenes Free radical reactions Selection rule for thermal and photochemical reactions. Frontier molecular orbital approach.	
SemIV			
CH-450 Chemistry of natural products	Introduction to secondary metabolism natural products, synthesis and application vitamins, enzymes,	Students understood importance of vitamins B1,B2,B6,B-12,Folic acid, C,D-1,E,K1 and K2, sources, structure, stereochemistry and biogenesis of vitamins, Role of enzymes in reaction	
CH-451 Synthetic methods in organic chemistry	Introduction to application of some elements in organic synthesis, Designing of organic synthesis, one and two stage	Students understood Transition metals in organic synthesis, Design the organic compounds, Role of Umpolung in organic synthesis, Polypeptide and poly nucleotides, principles of green chemistry solvents catalyst and	

and two stage

chemistry, solvents, catalyst and

	disjunction, protection of	reaction conditions.
	group, advanced	reaction conditions.
	0 1	
	synthetic reactions.	Contractor a la sector a 1 Constitución esta des
CH_452Heterocyclic	Introduction to	Students understood Synthetic routes,
chemistry, Chiron	heterocycle chemistry	reaction and reactivity of heterocyclic
approach, Chiral drugs	and chiral approach.	compounds, important terms used in
and medicinal chemistry	Drug discovry, synthesis	medicinal chemistry, structure of triose,
		pentose, hexose, stereochemistry and
		reactions. Synthesis and
		pharmacological activity of S-
		Lbuprofin, S- Metaprolol
CH-O2 Organic	Introduction to various	Students are able to separate organic
chemistry	types of organic	compound in different phases,
practical(Ternary	mixtures, their	Qualitative analysis of organic
mixture)	separation, identification	compounds, distillation techniques,
	and purification and	Detection of elements N, S, X,
	chromatographic study	Purification techniques.
CH-O-3 Organic	Introduction to organic	Students are able to perform three stage
chemistry Practical(Three	three stage preparation.	preparation, draw the reaction
stage preparation)	Purification and	mechanism, purify the organic
	chromatographic study	compounds by crystallization, perform
	of organic compounds	chromatographic technique to check
		completion of reaction, apply the
		knowledge about different reaction
		conditions.
CH-O4 –Short Research	Introduction to research,	Students understood literature survey
Project	survey literature review,	for the topic of the project, Standardize
110jeet	synthesis of raw	reaction conditions for synthesis, new
	products, purification	methods of synthesis, isolation of
	and analysis of products	product and give mechanism, handle
		instruments for analysis and discuss
		their experimental results, Used ICT
		tools to prepare project reports and
		present it using power point
		presentation, worm within a small team
		to achieve a common research goal.